

What Are SN1 and SN2 Substitution Reactions?

Worksheet

SN1 is a two-step mechanism through a carbocation intermediate (order-1 kinetics, favored by weak nucleophiles and polar solvents). SN2 is one-step, direct attack with inversion of configuration (order-2 kinetics, favored by strong nucleophiles and polar aprotic solvents).

Questions

1. Tertiary alkyl bromide + weak nucleophile + polar protic solvent ?

- A) SN2
- B) SN1
- C) E2
- D) No reaction

2. What does 'Walden inversion' mean?

- A) Carbocation forms
- B) Configuration flips during backside attack
- C) Solvent changes
- D) Nucleophile weakens

3. Primary alkyl halide + CN in DMF kinetics?

- A) First-order
- B) Second-order
- C) Zero-order
- D) Mixed

4. Which solvent favors SN2?

- A) Ethanol
- B) DMSO
- C) Water
- D) Acetic acid

5. Predict the mechanism for (CH₃)₂CH-Br with CN in DMSO.

6. What happens with CH₃-I and HO⁻ (protic solvent)?

7. Explain the stereochemistry when (R)-2-bromobutane reacts via SN2.

8. Define: What is SN1 and SN2?

9. Define: Which is faster: SN1 or SN2 with a strong nucleophile?

10. Define: Stereochemistry of SN2?

Answer Key

1. B) SN1 - SN1 is favored: bulky substrate (carbocation stability), weak Nu, polar protic solvent.
2. B) Configuration flips during backside attack - SN2 backside attack causes inversion of stereochemistry.
3. B) Second-order - SN2 is bimolecular rate = $k[\text{RX}][\text{Nu}]$ (second-order).
4. B) DMSO - DMSO is polar aprotic - does not solvate nucleophile as strongly, keeping it reactive.
5. Substrate: tertiary alkyl halide (favors carbocation) Nucleophile: CN (strong) Solvent: DMSO (polar aprotic) Conclusion: SN2 dominates due to strong Nu and polar aprotic solvent. Product: (CH)₃C-CN with inversion.
6. Substrate: primary alkyl halide Nucleophile: HO (weak in protic solvent) Solvent: HO (polar protic) Conclusion: SN2 is favored for primary. Product: CH-OH (methanol) with inversion.
7. SN2 requires backside attack. Walden inversion occurs. Configuration inverts from R to S. Product: (S)-2-nucleophile-butane.
8. SN1 = unimolecular substitution (2 steps, carbocation). SN2 = bimolecular (1 step, direct).
9. SN2 is faster - direct attack, single activation barrier.
10. Inversion of configuration (Walden inversion) due to backside attack.

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