

What is Density?

Worksheet

Density is mass per unit volume: $= m/V$, measured in kg/m or g/cm. Denser materials pack more mass into the same space.

$$\rho = \frac{m}{V}$$

Questions

1. A 200 g object has a volume of 50 cm. Its density is...
 - A) 4 g/cm
 - B) 0.25 g/cm
 - C) 10000 g/cm
 - D) 250 g/cm
2. Which of these floats on water (density 1 g/cm)?
 - A) Iron (7.87)
 - B) Gold (19.3)
 - C) Ice (0.92)
 - D) Lead (11.3)
3. What is the unit of density in SI?
 - A) N/m
 - B) kg/m
 - C) m/kg
 - D) kgm
4. A liquid has a density of 0.9 g/cm and a volume of 200 cm. Its mass is...
 - A) 180g
 - B) 222g
 - C) 90g
 - D) 1800g
5. A rock has a mass of 300 g and a volume of 100 cm. Find its density.
6. A metal block has a mass of 5400 kg and a volume of 2 m. Find its density.
7. An object has a density of 0.8 g/cm and a volume of 50 cm. Find its mass.
8. Define: What is density?
9. Define: Common density units?
10. Define: Why does an object float or sink?

Answer Key

1. A) $4 \text{ g/cm} = m/V = 200/50 = 4 \text{ g/cm}$.
2. C) Ice (0.92) - Ice is less dense than water, so it floats.
3. B) $\text{kg/m} - \text{Density} = \text{mass/volume kg/m}$.
4. A) $180\text{g} - m = V = 0.9 \cdot 200 = 180\text{g}$.
5. $= m/V = 300/100 = 3 \text{ g/cm}$
6. $= m/V = 5400/2 = 2700 \text{ kg/m}$
7. $m = V = 0.8 \cdot 50 = 40 \text{ g}$
8. Mass per unit volume: $= m/V$.
9. $\text{kg/m (SI) or g/cm (common in labs); } 1 \text{ g/cm} = 1000 \text{ kg/m}$.
10. It floats if its density is less than the fluid's density, and sinks if greater.

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