

What Is Nuclear Physics?

Worksheet

Nuclear physics is the study of atomic nuclei, covering radioactive decay, binding energy, and the fission and fusion reactions that release enormous amounts of energy.

$$E = \Delta m c^2$$

Questions

1. What is the half-life of a radioactive substance?

- A) The total time it takes to decay completely
- B) The time for half the sample to decay
- C) The time for the sample to double
- D) The energy released during decay

2. Which process splits a heavy nucleus into smaller ones?

- A) Fusion
- B) Fission
- C) Ionization
- D) Beta decay

3. What powers the sun?

- A) Nuclear fission
- B) Chemical combustion
- C) Nuclear fusion
- D) Radioactive decay alone

4. A sample with a 5-year half-life starts at 200 g. How much remains after 10 years?

- A) 100 g
- B) 50 g
- C) 25 g
- D) 150 g

5. A 100 g sample of a radioactive isotope has a half-life of 10 years. How much remains after 30 years?

6. Iodine-131 has a half-life of about 8 days. Starting with 40 g, how much is left after 24 days?

7. A sample decays from 80 g to 20 g in 12 years. What is its half-life?

8. Define: What is inside an atomic nucleus?

9. Define: What is radioactive half-life?

10. Define: What is nuclear fission?

Answer Key

1. B) The time for half the sample to decay - Half-life is defined as the time needed for half of a radioactive sample to decay.
2. B) Fission - Fission splits a heavy nucleus (like uranium-235) into lighter nuclei, releasing energy.
3. C) Nuclear fusion - The sun generates energy by fusing hydrogen nuclei into helium in its core.
4. B) 50 g - 10 years = 2 half-lives $200 (1/2) = 50$ g.
5. 30 years = 3 half-lives $N = 100 (1/2)^3 = 100 \cdot 0.125 = 12.5$ g
6. 24 days 8 days = 3 half-lives $N = 40 (1/2)^3 = 40 \cdot 0.125 = 5$ g
7. $20/80 = 1/4 = (1/2)^2$ 2 half-lives occurred 2 half-lives = 12 years $t = 12 / 2 = 6$ years
8. Protons and neutrons (collectively called nucleons), held together by the strong nuclear force.
9. The time it takes for half of a radioactive sample to decay.
10. The splitting of a heavy nucleus into smaller nuclei, releasing energy - used in nuclear reactors.

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